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**KONCHILIK SANOATIDA RUDALARNI GRAVITATSIYA USULIDA  
BOYITISH NAZARIYASI.**

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**Annotatsiya:** Gravitatsiya usuli bilan boyitishda ma’lum kattalikdagi xar xil zichlikka ega bo‘lgan zarrachalar muhitda (havoli, suvli) xarakat qiladilar. Zarrachalarning muhitda xarakatlanish qonuniyatlarini mexaniqa, gidravlika, fizika qonunlari xamda gidrodinamika va aerodinamikalarning asosiy nazariyalari orqali tushuntiriladi. Ba’zi bir xollarda statistik fizika va ehtimollar nazariyalaridan foydalanipladi.

**Kalit so‘zlar:** Gravitatsiya, turbulent, ekvivalent diametr, laminar, og‘ir suyuqlik, suspenziya.

**Kirish:** Gravitatsiya usuli bilan boyitish (saralash) asosan zarrachalarning og‘irlik kuchi ta’sirida ma’lum muhitda xar xil tezlik bilan xarakatlanishiga asoslangan.

Bu usul, boyitish usullari ichida eng qadimgisi hisoblanadi.

Odamzot birinchi bo‘lib daryo qumlaridan oltinni yuvib olishni o‘rganganlar. Bu esa gravitatsiya usuliga kiradi. Gravitatsion boyitishning prinsiplari (tamoyillari) odamlarga 2 ming yil oldin ma’lum bo‘lgan va birinchi bo‘lib Pliniy, keyinrok Agrikol tomonidan yozma ravishda bayon etilgan. Vaqt o‘tishi bilan bu usul takomillashib borgan, unumidorlikni oshirish uchun xar xil moslamalar, qurilmalar ixtiro qilingan. Eramizdan avvalgi V asrlardayok (Gerodot zamonida) Sochma konlarni boyitishda shlyuzlar, butaralar ishlatilgan.

Gravitatsion boyitishning eng rivojlanish davri XIX asr oxiri va XX asrning o‘rtalariga to‘g‘ri keladi. Bu davrda hozirgi zamon gravitatsiya usulini nazariyasi shakllanadi, juda ko‘p yangi apparatlar yaratiladi. Hozirgi vaqtida gravitatsiya usuli dunyoning barcha boyitish fabrikalarida foydalanipladi. Bunga sabab, usulning soddaligi, kamxarjligi, unumidorligining yuqoriligidir.

**Qattik jism va muhitning xossalari.**

Hozirgi vaqtida tabiatda uch mingga yaqin mineral borligi ma’lum bo‘lib, shulardan 250 tasi gravitatsiya usuli bilan boyitish jarayonida qatnashadi. Yuqorida aytiganidek, gravitatsiya zarrachalarning zichligi, shakli va o‘lchamlari turli bo‘lib, muhitda xar xil tezlik bilan xarakatlanishi qonuniyatiga asoslangan boyitish usulidir.

Shuning uchun, zichlik, shakl, zarracha o‘lchami va muhit xossalari degan tushunchalar bilan tanishib chiqamiz.

**Qattik zarracha xossalari.**

Zarrachaning zichligi deb, uning massasini ( $m$ ) hajmiga nisbatiga aytildi va  $\delta$  - xarfi bilan belgilanadi:

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$$\delta = \frac{m}{V}, \text{ kg/m}^3 \quad (1)$$

Bu yerda: m - zarrachaning massasi, kg;  
V - zarrachaning xajmi, m<sup>3</sup>.

Zarrachalarning kattaligi uning geometrik o'lchami bilan belgilanadi. Hisob-kitoblarni soddalashtirish uchun zarra shaklini inobatga olinmagan xolda, «ekvivalent diametr» degan tushuncha kiritilgan:

$$V = \frac{\pi d^2}{6} \quad (2)$$

$$d_y = \sqrt[3]{\frac{6V}{\pi}} = \sqrt[3]{\frac{6m}{\pi\delta}} = 1.24 \sqrt[3]{\frac{m}{\delta}} \quad (3)$$

Minerallarning zichliklari 1300-19300 kg/m<sup>3</sup> atrofida bo'ldi (1-jadval).

1-jadval.

Minerallarning zichliklari

Mineral	Kimyoviy ifodasi	Zichligi, kg/m <sup>3</sup>	Foydali modda	Foydali moddaning miqdori, %
Oltin	Au	19,300	Oltin	100
Galenit	PbS	7400-7600	Ko'rg'oshin	86,6
Molibdenit	MoS <sub>2</sub>	4300-5000	Molibden	59,94
Magnetit	Fe <sub>3</sub> S <sub>2</sub>	4500-5300	Temir	72,4
Pirit	FeS <sub>2</sub>	4900-5200	Temir	46,55
Kvars	SiO <sub>2</sub>	2650	Kremniy	46,7
Kalsiy	CaCO <sub>3</sub>	2700	SaO	56
Magnezit	MgCO <sub>3</sub>	3000	MgO	47,6
Ko'mir	C	1300-1800	Uglerod	-

Zarra shakli xar xil bo'lib, u mineralning tabiatiga bog'liq. Xisoblashlar uchun shakl koeffitsienti degan tushuncha kiritilgan:

$$\varphi = \frac{Sm}{S} = \frac{4,87V^{2/3}}{S} \quad (4)$$

Bu yerda:  $\varphi$  - shakl koeffitsienti

Sm - sharning yuzasi

S - noto'g'ri shaklli zarracha yuzasi

Quyida shakl koeffitsientining qiymati keltirilgan.

Sharsimon - - - - - 1,0

Yumaloqroq - - - - - 0,8 -0,9

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Burchakli - - - - - 0,7-0,8

Yalpoq - - - - - 0,6-0,7

### Muhitning xossalari

Gravitatsiya usuli bilan boyitishda muhit vazifasini havo, suv, og‘ir suyuqlik, suspenziya (loyqa) elektr yoki magnit maydoniga joylashtirilgan suyuq elektrolit o‘tashi mumkin. Boyitishda muhitning quyidagi xossalari axamiyatga molik: Zichligi, qovushqoqligi, xarakatlanayotgan jismga qarshiligi va barqarorligi.

Yuqorida keltirilgan muhit xossalari unda ajraliyotgan zarrachalar xarakatiga katta ta’sir qiladi. Shuning uchun ularning qiymatlari boyitish dastgoxlarini xisoblashda inobatga olinadi.

Muhitning zichligi deb, muhit massasini, uning xajmiga nisbatiga aytiladi.

$$\Delta = \frac{m}{V}, \text{ kg/m}^3 \quad (5)$$

Ma’lumki, og‘irlilik bilan massa o‘zaro m g=G nisbatiga ega. Tenglamani ikala qismini hajmiga bo‘lsak:

$$\Delta g = \rho \quad (6)$$

Ifodani olamiz. Bu zichlik bilan solishtirma og‘irlilik o‘zaro bog‘liq ekanligini ko‘rsatadi.

Bu yerda: G - muxitning og‘irligi, kg

g - erkin tushish tezlanishi, 9,81 m/s<sup>2</sup>

$\rho$  – muxitning solishtirma og‘irligi, kg/m<sup>3</sup>

Suvning zichligi 20<sup>0</sup> S haroratda 1000 kg/m, havoning zichligi-1,23 kg/m<sup>3</sup> ga teng.

1.2- jadvalda boyitish jarayonida ishlatiladigan ba’zi bir og‘ir suyuqliklari tavsifi keltirilgan.

Og‘ir suyuqliklardan ZnCl<sub>2</sub>, CaCl<sub>2</sub>, bromofrom, Tula va klerchi suyuqliklari sanoatda ko‘prok qo‘llaniladi. Og‘ir suyuqliklarni (og‘irlashtirgichlarni) eritmadagi konsentratsiyalarni o‘zgartirib muxit zichligini o‘zgartirish mumkin, masalan:

Bromofrom miqdori, hajmiy foiz hisobida	100	75	50	25
Eritmaning zichligi	2890	2430	1890	1320

2-jadval.

#### Og‘irlilik suyuqliklari tavsifi

Suyuqlik	Kimyoviy ifodasi	Zichligi, kg/m <sup>3</sup>	Suvdagi eruvchanligi
Sink xloridi	ZnCl <sub>2</sub>	2500	Eriydi
Kalsiy xloridi	CaCl <sub>2</sub>	2500	Eriydi
Bromofrom	CHBr <sub>3</sub>	2890	Eriydi
Tula suyuqligi	CH <sub>2</sub> (COOTI) <sub>2</sub>	3170	Eriydi

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Klerchi suyuqligi	HCOOTI	4250	Eriydi
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Suspenziya (loyqa) ning zichligi og‘irlashtirgichning zichligi va uning suspenziyadagi xajmiy miqdoriga bog‘liq. Bu kattaliklarni o‘zaro bog‘liqligini quyidagi mulohaza yuritish bilan aniqlash mumkin.

Begilarni qo‘yamiz:

$$V_c - \text{suspneziyaning xajmi}, 1 \text{ m}^3$$

$$\Delta c - \text{suspenziyaning zichligi}, \text{kg/m}^3$$

$$V_T - \text{suspenziyayadagi og‘irlashtirgichni xajmiy miqdori}$$

$$\delta - \text{og‘irlashtirgichni zichligi}, \text{kg/m}^3$$

$$V_c = V_T = V \text{ suv-suspenziyadagi suvning xajmi}$$

$$\Delta - \text{suvning zichligi}, 1000 \text{ kg/m}^3$$

Quyidagi tenglamani yozishimiz mumkin:

$$V_c \Delta c = V_T \delta + (V_c - V_T) \Delta c y\sigma$$

bundan

$$\Delta c = V_T \delta + (1 - V_T) 1000$$

$$V_T = \frac{\Delta c - 1000}{\delta_T - 1000} \quad (7)$$

Xajm birligida suspenziyadagi og‘irlashtirgichning miqdori

$$G = V_\delta \delta = \frac{\Delta - 1000}{\delta_m - 1000} \delta \quad (8)$$

Suspenziyadagi og‘irlashtirgichni massa bo‘yicha konsentratsiyasi

$$C_\delta = V_\delta \frac{\delta}{\Delta \tilde{n}} = \frac{\Delta \tilde{n} - 1000}{\delta - 1000} \cdot \frac{\delta}{\Delta \tilde{n}} \quad (9)$$

Berilgan xajmda suspenziya tayyorlash uchun kerak bo‘lgan og‘irlashtirgichni miqdori:

$$Q = V_c V_T \delta = V \frac{\Delta c - 1000}{\delta - 1000} \delta \quad (10)$$

Amalda ko‘mirni boyitish uchun tayyorlashda magnetit boyitmasi ( $\delta_t = 4400$ ), rudalarni boyitish uchun esa ferrosilitsiy ( $\delta_t = 6800$ ) ishlatiladi.

Suyuq muxitni zichligini oshirish uchun eletr tokidan foydalanilsa xam bo‘ladi. Chunki, magnit va eletr maydonlariga o‘rnatilgan elektrolitning zichligi magnit va elektr tokining parametrlariga bog‘liq bo‘ladi. Buning uchun magnitogidrodinamik qurilmasidan foydalaniлади.

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Zichliklari xar xil bo‘lgan zarrachalarni bir-biridan ajratish uchun amalda «MGD-separatsiya» qurilmasidan foydalilanildi (4-rasm).

Qurilma dielektrik materialdan yasalgan N va S qutbli idish (1), elektr tokiga ulangan metall elektrod (2,3) lardan iborat.

Elektrodlar tokka ulanganda elektrolit orqali tok o‘ta boshlaydi va magnit maydoni bilan o‘zaro ta’sirlanadi. Bu kuch quyidagicha ifodalanadi:

F=VII

(11)

Bu yerda: V – qutblar oralig‘idagi magnit maydonini induksiyasi;

I – elektrolit orqali o‘tadigan tok

I – elektrodlar orasidagi masofa.

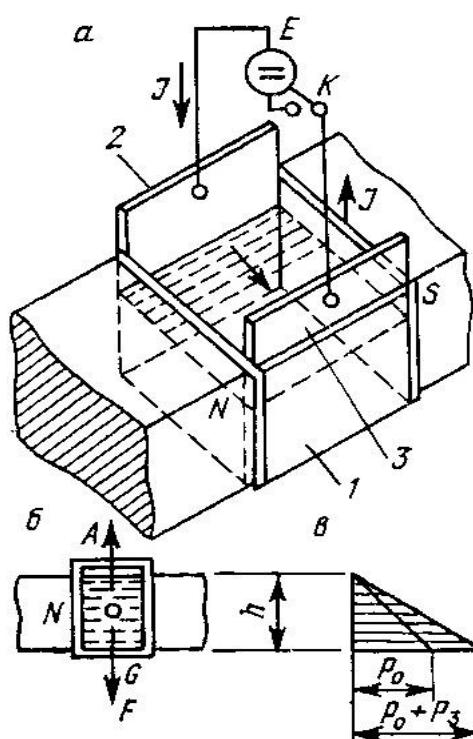
Elektrolitning elementar xajmiga ta’sir qilayotgan elektromagnit kuchi, pastga yo‘nalgan:

dF=Bilds

(12)

Bu yerda: i – elektrolitdagi tok zichligi;

Ds – elektrolitni elementar qirqimi;



4 -rasm. Gidrodinamik qurilma: a-qurilma sxemasi; b-mineral zarrachaga ta’sir qilayotgan kuch sxemasi; v- balandligi bo‘yicha elektrolit qatlamiga ta’sir etayotgan kuch dinamikasi; A- itaruvchi kuch; V – og‘irlik kuchi; F-elektromagnit

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